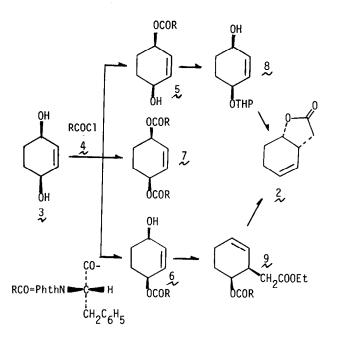
## SYNTHESIS OF AN OPTICALLY PURE PROSTAGLANDIN INTERMEDIATE FROM CIS-2-CYCLOHEXENE-1,4-DIOL

Shiro Terashima,<sup>a)\*</sup> Munehiko Nara,<sup>b)</sup> and Shun-ichi Yamada<sup>C)</sup> Faculty of Pharmaceutical Sciences, University of Tokyo, Hongo, Bunkyo-ku, Tokyo 113, Japan;<sup>a)</sup> Tokyo Research Laboratories, Kowa Co. Ltd., 2-17-43, Noguchicho, Higashimurayama, Tokyo 189, Japan;<sup>b)</sup> Faculty of Pharmaceutical Sciences, Josai University, 1-1, Keyakidai, Sakado-shi, Saitama 350-02, Japan<sup>C)</sup>

(Received in Japan 13 February 1978; received in UK for publication 7 March 1978) As an efficient method for producing optically active compound, the use of symmetrically functionalized <u>meso</u>-compound as a resolution substrate is considered most desirable because it is theoretically possible to utilize the total amount of <u>meso</u>-compound for synthesizing one requisite enantiomer usable for synthetic scheme.<sup>1,2)</sup> Although this concept has been realized by preparing optically pure lactone(1) from <u>cis</u>-2-cyclopentene-1,4-diol,<sup>1)</sup> preparation of another optically pure prostaglandin intermediate(2)<sup>3)</sup> is attempted to explore the generality of this novel methodology, by using <u>cis</u>-2-cyclohexene-1,4-diol(3)<sup>4</sup> as a resolution substrate and (S)-N-phthaloylphenylalanyl chloride(4)<sup>5)</sup> as a chiral compound.

As shown in the scheme, acylation of 3 with  $4(1.5 \text{ eq})(\text{anhyd. KHCO}_3(10 \text{ eq})$  in THF, rt, 3.5 days), followed by extraction with ether and separation by a silica gel column(CHCl<sub>3</sub>), gave a mixture of the monoesters(5 and 6)<sup>6b</sup> as an oil(54%),  $[\alpha]_D^{20}$ -124°(c=1.8, CHCl<sub>3</sub>), and the crystalline diester(7)<sup>6)</sup>(7.8%), mp 182.5-183°C,  $[\alpha]_D^{20}$ -154°(c=1.2, CHCl<sub>3</sub>). The mixture of 5 and 6 was dissolved in ether, and the ethereal solution was cooled in an ice bath. This operation yielded a mixture of 5 and 6 as colorless needles<sup>6b</sup>(34% from 3) in which 5 was predominant, mp 94-98°C,  $[\alpha]_D^{20}$ -106°(c=1.5, CHCl<sub>3</sub>). Further recrystallization of this substance from ether afforded a mixture of 5 and 6(ca.3:1)<sup>7,8)</sup> as colorless needles<sup>6b</sup>(17% from 3), mp 97.5-100°C,  $[\alpha]_D^{20}$ -75.3°(c=1.1, CHCl<sub>3</sub>). Concentration of the original ethereal mother liquor gave a mixture of 5 and 6(ca. 1:5)<sup>7</sup>) as colorless needles<sup>6b</sup>(19% from 3), mp 69-71°C,  $[\alpha]_D^{20}$ -162°(c=1.3, CHCl<sub>3</sub>).

Protection of the alcoholic function of the mixture of 5 and 6(ca. 3:1) as tetrahydropyranyl(THP) ether, followed by hydrolysis of the chiral acyl group(KOH(2.0 eq)-aq. MeOH, rt, 5 hr), gave oily  $g^{6b}$  in a quantitative yield,  $[\alpha]_D^{20}$ -17.3°(c=1.2, CHCl<sub>3</sub>). The alcohol(8) was submitted to Claisen rearrangement(triethyl orthoacetate-hydroquinone (catalytic amount), 160°C, 24 hr), and the rearrangement product was hydrolyzed(KOH(2.0 eq)-aq. MeOH), then simultaneously deprotected and lactonized(aq. AcOH, rt, 2 days), giving the desired lactone(2) as a semisolid<sup>6b</sup>(79% from 8), bp 96-97°C(4 mmHg),  $[\alpha]_D^{20}$ -15.0°(c=1.3,



MeOH), 50% optically pure.<sup>9)</sup> Two repeated recrystallizations from ether-hexane gave optically pure 2 as colorless prisms<sup>6)</sup>(53% from 8), mp 68.5-69.5°C,  $[\alpha]_D^{20}$ -29.8°(c=0.9, MeOH).

On the other hand, when a mixture of 5 and 6(ca. 1:5) was directly treated under the condition of Claisen rearrangement similar to that for 8, and the rearrangement product(9) was successively hydrolyzed(KOH(4.0 eq)-aq. MeOH, rt, 5 hr) and lactonized(HCl(catalytic amount), rt, 24 hr), the lactone(2),  $^{6b}$  [ $\alpha$ ]<sup>20</sup> -20.2°(c=1.2, MeOH), 67% optically pure, <sup>9)</sup> was obtained as a colorless semisolid(87% from 6). Recrystal-

lization from ether-hexane readily gave optically pure 2 as colorless prisms  $^{6b}$  (53% from 6), mp 68-69.5°C,  $[\alpha]_D^{20}$ -30.0°(c=1.0, MeOH).

As described above, the preparation of optically pure 2 could be accomplished in 15% overall yield from 3. Further studies on applicability of the novel method to preparation of optically pure key intermediate for natural product synthesis are under progress in these laboratories.

## References and Notes

- 1) S. Terashima, S. Yamada, and M. Nara, Tetrahedron Letters, 1977, 1001.
- 2) For a detailed discussion on the concept, see ref. 1.
- a) E.J. Corey and T. Ravindranathan, Tetrahedron Letters, <u>1971</u>, 4753. b) E.J. Corey and B.B. Snider, <u>Ibid.</u>, <u>1973</u>, 3091. c) <u>Idem.</u>, J. Org. Chem., <u>39</u>, 256(1974).
- 4) C. Kaneko, A. Sugimoto, and S. Tanaka, Synthesis, <u>1974</u>, 876.
- 5) J.C. Sheehan, D.W. Chapman, and R.W. Roth, J. Am. Chem. Soc., 74, 3822(1952).
- 6) Satisfactory a) analytical and b) infrared and nuclear magnetic resonance data have been obtained for this compound.
- 7) This was calculated from the optical purity of 2 derived from this compound.
- 8) Although repeated recrystallizations increased the ratio of 5 to 6 upto 94:6, effort to prepare completely pure 5 seemed useless because it was found that partially optically active 2 could readily give optically pure sample when recrystallized from ether-hexane.
- 9) The lactone(2) showing  $[\alpha]_D^{20}$ -30.0°(c=1.0, MeOH), was assumed to be optically pure(lit., <sup>3c)</sup>  $[\alpha]_D^{27}$ -28°(c=0.83, MeOH)).